

Chemical Modulation of the Chaperone Function of Human α A-Crystallin

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Received December 22, 2007; accepted March 3, 2008; published online March 15, 2008

α A-crystallin is abundant in the lens of the eye and acts as a molecular chaperone by preventing aggregation of denaturing proteins. We previously found that chemical modification of the guanidino group of selected arginine residues by a metabolic α -dicarbonyl compound, methylglyoxal (MGO), makes human α A-crystallin a better chaperone. Here, we examined how the introduction of additional guanidino groups and modification by MGO influence the structure and chaperone function of α A-crystallin. α A-crystallin lysine residues were converted to homoarginine by guanidination with *o*-methylisourea (OMIU) and then modified with MGO. LC-ESI-mass spectrometry identified homoargpyrimidine and homohydroimidazolone adducts after OMIU and MGO treatment. Treatment with 0.25 M OMIU abolished most of the chaperone function. However, subsequent treatment with 1.0 mM MGO not only restored the chaperone function but increased it by ~40% and ~60% beyond that of unmodified α A-crystallin, as measured with citrate synthase and insulin aggregation assays, respectively. OMIU treatment reduced the surface hydrophobicity but after MGO treatment, it was ~39% higher than control. FRET analysis revealed that α A-crystallin subunit exchange rate was markedly retarded by OMIU modification, but was enhanced after MGO modification. These results indicate a pattern of loss and gain of chaperone function within the same protein that is associated with introduction of guanidino groups and their neutralization. These findings support our hypothesis that positively charged guanidino group on arginine residues keeps the chaperone function of α A-crystallin in check and that a metabolic α -dicarbonyl compound neutralizes this charge to restore and enhance chaperone function.

Key words: α A-crystallin, chaperone, homoarginine, homoargpyrimidine, homohydroimidazolone.

Abbreviations: sHsps, small heat shock proteins; DTT, dithiothreitol; CS, citrate synthase; OMIU, *o*-methylisourea; MGO, methylglyoxal; TNS, 2-(*p*-toluidino) naphthalene-6-sulphonic acid, sodium salt; FRET, fluorescence resonance energy transfer; LC-ESI-MS, liquid chromatography-electrospray ionization mass spectrometry; AIAS, 4-acetamido-4'-((iodoacetyl)amino)-stilbene-2,2'-disulphonic acid; LYI, lucifer yellow iodoacetamide.

Alpha-crystallin is a major structural protein in the lens of the eye. It is a polydisperse protein, usually occurring in aggregates of 40 subunits with a molecular weight ~800 kDa (1). The polymeric aggregate consists of both α A- and α B-crystallins, usually in a stable ratio of 3:1 (1). The α A- and α B-crystallins both belong to a family of proteins known as small heat-shock proteins, or sHsps, which includes Hsp27 (2, 3). The sHsps have three distinct structural domains. The inner core domain, or α -crystallin domain, is about 80 amino acids in length (4). This is flanked by an N-terminal domain that varies in length and sequence. The C-terminal domain is flexible without a rigid

structure and is believed to be responsible for the solubility of many sHsps, including α -crystallin (5).

Although sHsps are resident 'housekeeping' proteins, they are synthesized in higher amounts in response to thermal, oxidative and other stresses in order to prevent cell damage (6, 7). Their molecular chaperone activity is one mechanism through which sHsps provide defence; in addition, they are anti-apoptotic (6–8). The chaperone function of α -crystallin was first demonstrated by Horwitz (9) and subsequently confirmed by a number of other investigators (10–12). The chaperone function enables α -crystallin to inhibit aggregation of denaturing proteins, and alteration of this process has been linked to certain diseases characterized by protein aggregates, including cataract formation and Alzheimer's disease (13, 14). Experimental evidence from congenital dominant mutations of α -crystallin (15–17) or from post-translational modifications (18–20) indicates that compromised chaperone function can lead to cataract formation.

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The hydrophobic regions of α -crystallin are thought to be sites of chaperone–substrate interaction that are essential for chaperone function (21–23). Several of these substrate binding sites have been identified (21–24), notably at amino acids 12–21 and 71–78 in α A-crystallin and 9–20, 28–34, 43–58 and 75–82 in α B-crystallin. Recent studies suggest that arginine residues, either within the protein interaction sites or elsewhere within the protein, dictate the chaperone function of both α A- and α B-crystallins (15, 25–27). Thampi and Abraham (28) found that cleavage of 11 C-terminal residues of rat α A-crystallin, including Arg-163, decreased chaperone function. The R116C mutation in α A-crystallin cripples its chaperone activity and promotes cataract formation in the human lens (15, 29) as does over-expression of R116C in the mouse lens (30). An R120G mutation in α B-crystallin similarly compromises the chaperone function and leads to cataract formation (25). Other mutations, such as R49C, also appear to be linked to cataract formation (31).

Amid these loss-of-chaperone effects relating to arginine residues, several studies suggest that a gain of such function is associated with deletion or modification of arginine residues in α -crystallin. Pasta *et al.* (32) showed that deletion of the ${}_{20}\text{SRLFDQFFG}_{28}$ sequence, and thus R21, in α A-crystallin made it a better chaperone. More recently, deletion of ${}_{54}\text{FLRAPS WF}_{61}$ sequence was reported to improve chaperone function (33). Our laboratory demonstrated that replacement or modification of specific arginine residues with methylglyoxal (MGO), a metabolic α -dicarbonyl, improved chaperone function of α A-crystallin, presumably by modification of R21, R49 and R103 to argpyrimidine (34). We observed similar but lesser effects with other arginine-modifying compounds, such as, phenylglyoxal, 1,2-cyclohexanedione and 2,3-butanedione, which confirmed that modification of guanidino group of arginine residues is responsible for the increased chaperone function of α -crystallin (34). Recently, we reported that replacement of R21 and R103 with alanine by

site-directed mutagenesis was equally effective (26) and that deletion of R56 in human α B-crystallin enhances its chaperone function by at least 35% as compared to the native protein (27). Our studies also confirmed that the positive charge on selected arginine residues modulates the chaperone function of α -crystallin.

Realization that MGO modification neutralized positive charge on selected arginine residues and enhanced the chaperone function of α A-crystallin prompted us to investigate if we could modulate the chaperone function by introducing additional arginine residues and their modification with MGO. Arginine residues could be introduced by cloning strategies at random places on the protein, but that might change the net charge on the protein and might disrupt its folding pattern as well. Another method is to replace positively charged lysine residues with arginine by site-directed mutagenesis. Since there are seven lysine residues, to replace them one at a time or in combination would require many mutations and therefore could be extremely tedious. The third strategy, the one we adopted in this study, is to introduce homoarginine (similar to arginine in that it carries positive charge on the guanidino group) residues by converting lysine residues to homoarginine by reacting with *o*-methylisourea (OMIU) (Fig. 1). This approach is simple and straightforward. This method yielded some surprising results: while modification by OMIU almost completely abolished the chaperone function, additional modification by MGO not only restored it, but it made α A-crystallin an even better chaperone than the unmodified protein. We then sought to define the structural changes associated with such dramatic changes in the chaperone function.

MATERIALS AND METHODS

Bovine insulin, citrate synthase (CS), dithiothreitol (DTT) and *o*-methylisourea (OMIU) were obtained from

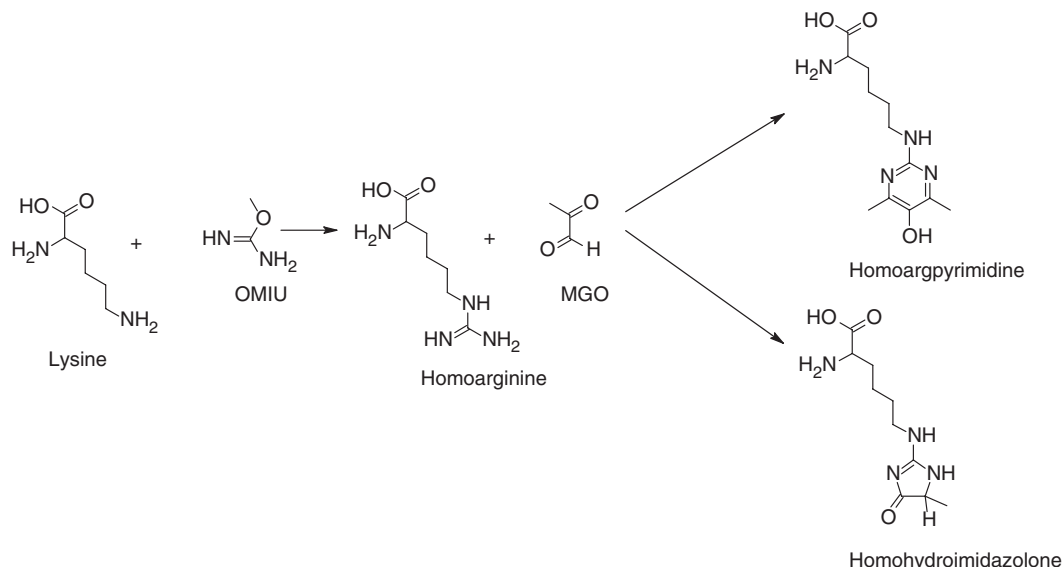


Fig. 1. Formation of homoarginine+MGO adducts on α A-crystallin. Lysine residues are converted to homoarginine by reaction with OMIU. Reaction of homoarginines with

produces homoargpyrimidine (HAP) and homohydroimidazolone (HHI) adducts on α A-crystallin.

Sigma Chemical Co., St. Louis, MO, USA. Citrate synthase was dialysed against 0.04 M HEPES buffer (pH 7.4) for 24 h before use. 2-*p*-toluidinylnaphthalene-6-sulphonate (TNS), lucifer yellow iodoacetamide (LYI) and 4-acetamido-4'-((iodoacetyl)amino)-stilbene-2,2'-disulphonic acid (AIAS) were obtained from Molecular Probes (Invitrogen, Carlsbad, CA, USA). All other chemicals were of analytical grade.

Cloning, Expression and Purification of Recombinant α A-Crystallin—The cDNA encoding human α A-crystallin was kindly provided by J. Mark Petrash, Washington University, St. Louis, MO, USA. Cloning, expression, and purification of this protein were done as previously described (26).

Chemical Modification of α A-Crystallin—Modification by OMIU

An aqueous solution of recombinant human α A-crystallin (1.0 mg/ml) was incubated with various concentrations (0 to 1 M) of OMIU at pH 10.5 for 48 h at 4°C as previously described (35). We then dialysed all solutions against 50 mM sodium phosphate buffer (pH 7.4) for 48 h to eliminate excess OMIU.

Modification by MGO

Unmodified and OMIU-modified α A-crystallin (0.5 mg/ml) in 0.05 M sodium phosphate buffer (pH 7.4) were incubated with 1 mM MGO for 2 days at 37°C. We then dialysed all solutions against 50 mM sodium phosphate buffer (pH 7.4) to eliminate any unreacted MGO.

BSA was similarly modified with OMIU (0.25 and 1.0 M) and MGO (1.0 mM) and dialysed.

Assessment of Chaperone Function—Chaperone activity of unmodified and modified α A-crystallin (20 μ g each) was measured in 96-microwell plates using a microplate reader (Molecular Devices, Model 190, Sunnyvale, CA, USA). Insulin, 80 μ g (0.32 mg/ml), was reduced by freshly prepared DTT (final concentration 20 mM) to break the inter-chain S–S bond; this results in aggregation of the B-chain. Aggregation was measured (at 25°C) in absence and presence of 20 μ g (0.08 mg/ml) unmodified and modified α A-crystallin by monitoring light scattering at 400 nm for 1 h (36). Citrate synthase, 15 μ g (0.06 mg/ml) protein in 0.04 M HEPES buffer, pH 7.4) was heated at 43°C in the presence or absence of 0.75 μ g unmodified and modified α A-crystallin, and light scattering was monitored at 360 nm (37). Reactions for both assays were carried out in a total volume of 250 μ l.

Amino Acid Analysis—Amino acid analyses were done at the Protein Chemistry Laboratory, Department of Biochemistry and Biophysics, Texas A&M University, College Station, TX, USA. Briefly, unmodified and modified α A-crystallin (treated with 0.1 to 1.0 M OMIU) were hydrolysed with 6 N HCl at 110°C for 20 h. The hydrolysed samples were evaporated to dryness and suspended in 300 μ l of Milli-Q water. For the analysis, 20 μ l of diluted sample was mixed with 250 μ l of 0.4 N borate buffer, and the pH was adjusted to 10.0. Five nanomoles of internal standard was added to all samples, standards and blanks. The assay was calibrated with two 5 nM standards that were not hydrolysed. All samples were analysed on a Hewlett Packard

AminoQuant System, which includes automated pre-column derivitization of the hydrolysed primary amino acids with *o*-phthalaldehyde (OPA) and the secondary amino acids with 9-fluoromethyl-chloroformate (Fmoc). Derivatized amino acids were separated by reverse phase HPLC on an HP 1090L and detected by photodiode array (UV-DAD).

Synthesis of Homoargpyrimidine—*t*-BOC-homoarginine (0.5 M) and MGO (1.0 M) were dissolved in 5.0 ml 0.2 M sodium phosphate buffer (pH adjusted to 7.4 with 10 N NaOH) and incubated at 50°C for 24 h. The incubation mixture (500 μ l) was injected onto a C₁₈ reverse phase semi-preparative column (Vydac, 218TP1010) with a water–acetonitrile gradient system. Solvent A was water with 0.1% trifluoroacetic acid (TFA), and solvent B was 50% acetonitrile in water with 0.1% TFA. The linear gradient program was as follows: 0 to 10 min, 0% B; 10 to 20 min, 30% B; 20 to 35 min, 50% B; 35 to 45 min, 100% B; 45 to 52 min, 100% B; 52 to 63 min, 0% B. The flow rate was 2.0 ml/min. The column effluent was monitored for fluorescence at 385 nm (excitation, 335 nm) with an online fluorescence detector. The fluorescent peak at R_t ~45 min was collected from 10 injections, dried in a Speed Vac concentrator, suspended in 0.5 N HCl and incubated at 50°C for 2 h to remove the *t*-BOC group. The sample was dried, suspended in 500- μ l solvent A and subjected to HPLC as described above. We noted a single homogenous peak at R_t ~28 min, which we collected and lyophilized. The yield was 12 mg. ¹H-NMR analysis of this product showed the following signals (CD₃OD): δ 3.94 (*t*, 1H, *J*=6.0 Hz), 3.48 (*t*, 2H, *J*=7.2 Hz), 2.44 (*s*, 6H), 1.97 (*m*, 2H), 1.45 to 1.75 (2H). ESI-MS, *m/z* 269 (M^+ +1). These characteristics are fully compatible with homoargpyrimidine [*N*⁶-(5-hydroxy-4,6-dimethylpyrimidin-2-yl)-lysine] (Fig. 1).

HPLC Assay for Argpyrimidine and Homoargpyrimidine—Protein samples (300 μ g each) were hydrolysed in 6 N HCl at 110°C for 20 h. The acid was evaporated in a Speed Vac system, and the pellet was suspended in 200 μ l water and filtered through a 0.45 μ m centrifugal filter. The amino acid content of each hydrolysate was estimated with ninhydrin as described (38). The samples were injected onto a C₁₈ reverse phase column (GraceVydac, 218TP54) and separated in a gradient system of water and acetonitrile. Solvent A was water with 0.01 M heptafluorobutyric acid (HFBA), and solvent B was 70% acetonitrile in water with 0.01 M HFBA. The solvent program was as follows: 0–39 min, 16% B; 40–50 min, 20% B; 50–60 min, 22% B; 60–62 min, 28% B; 62–71 min, 100% B; 71–80 min, 16% B. We monitored the column eluate with an online fluorescence detector set at excitation/emission wavelengths of 335/385 nm. Under these conditions, argpyrimidine had an R_t of ~28 min, and homoargpyrimidine had an R_t ~45 min. We estimated amounts of argpyrimidine and homoargpyrimidine in our protein samples by comparison with peak areas of synthetic standards.

HPLC-ESI-Mass Spectrometry—Protein reduction, S-alkylation and digestion

Unmodified α A-crystallin (20 μ g), α A-crystallin modified by OMIU (0.25 M) or by the combination of 0.25 M OMIU + 1 mM MGO were reduced with DTT for 2 h at

30°C in 20 µl of 1 M Tris–HCl (pH 8.0) containing 8 M urea. The samples were S-alkylated by treating with 2.5 mM iodoacetamide for 30 min at 25°C in the dark. Unreacted reagents were removed by dialysis of the proteins against 0.1% formic acid, and then the samples were lyophilized, re-dissolved in 1 M Tris–HCl (pH 8.0) and digested by the endoproteinase, Asp-N (Roche Applied Science, Indianapolis, IN, USA) for 16 h at 25°C. The enzyme-to-protein ratio was 1:100 (w/w).

Identification of homoarginine, homohydroimidazolone and homo-argpyrimidine

LC–ESI–MS/MS analyses of the Asp-N digests were done with a Q-Star XL quadrupole/time of flight (TOF) mass spectrometer (Applied Biosystems–MDS Sciex, Foster City, CA, USA) coupled to an Agilent 1100 capillary HPLC system (Agilent, Santa Clara, CA, USA) as described previously (39). The data were manually analysed by examining the spectra for non-modified Asp-N peptides as well as for ions with the mass increment predicted from our chemical modifications of the protein (a mass increase of 42 Da for homoarginine, 96 Da for homohydroimidazolone and 122 Da for homo-argpyrimidine). The peptide sequences were confirmed by their product ion spectra.

Tryptophan Fluorescence Measurements—Tryptophan fluorescence of α A-crystallin solutions was measured in a LS-55 Perkin Elmer spectrofluorometer as described previously (26). The excitation wavelength was set to 295 nm, and emission spectra were recorded between 310 and 400 nm. Excitation and emission band passes were 5 nm each. Data were collected at a 0.5 nm wavelength resolution.

Circular Dichroism (CD) Spectroscopy—Far- and near-UV CD spectra were recorded at 25°C in a Jasco 810 spectropolarimeter (Jasco, Inc., Japan). The spectra were measured using 1 and 10 mm cells with 0.2 mg/ml protein for far- and 1.0 mg/ml for near-UV CD. Our reported CD spectra comprise an average of five scans, and we used the curve-fitting program CONTINLL to analyse the secondary structure of unmodified, OMIU and OMIU + MGO modified α A-crystallin (40).

TNS Fluorescence Measurements— α A-Crystallin (0.1 mg/ml) was incubated in a methanolic solution of TNS (100 µM) for 2 h at 25°C. Fluorescence of TNS-bound samples was measured between 350 and 520 nm following excitation at 320 nm and recorded at 25°C using a LS-55 Perkin Elmer spectrofluorometer with the excitation and emission band passes set at 5 nm. Data were collected at a 0.5 nm resolution.

Fluorescence Labelling of Recombinant α A-Crystallin with LYI and AIAS—The cysteine residue at position 131 in α A-crystallin was labelled separately with the fluorescent probes AIAS and LYI as described (41). The covalently labelled α A-crystallin was then separated by passage through Sephadex G-25 column (20 × 2.0 cm) equilibrated with buffer (pH 7.5) containing 100 mM NaCl, 2 mM DTT and 50 mM sodium phosphate. The first fluorescence peak, which contained the labelled proteins, was collected and dialysed for 24 h against 50 mM phosphate buffer (pH 7.5). We followed the same

procedure to calculate the percentage labelling of AIAS/LYI in α A-crystallin as described in (41).

Measurements of Subunit Exchange Rate—The excitation maxima of AIAS- and LYI-labelled α A-crystallin were found to be 335 nm and 435 nm; the emission maximum of AIAS-labelled protein was 415 nm and that for the LYI-labelled α A-crystallin was 525 nm. Subunit exchange kinetics was measured according to a previously published method (42). Subunit exchange was accomplished by mixing equal volumes of 0.4 mg/ml AIAS-labelled α A-crystallin (unmodified and modified) and 0.4 mg/ml LYI-labelled α A-crystallin (unmodified and modified) at 37°C in 50 mM phosphate buffer (pH 7.5) containing 2 mM DTT and 100 mM NaCl. At various time intervals, 20 µl of the reaction mixture was removed and diluted 100-fold with the same buffer. The fluorescence spectra of the samples were recorded from 360 to 600 nm at room temperature with an LS-55 Perkin Elmer spectrofluorometer after excitation at 335 nm. Both excitation and emission monochromators had a slit width of 5 nm. We measured the intensity at 415 nm and calculated the subunit exchange rate from the following equation:

$$F(t)/F(0) = D_1 + D_2 e^{-kt}$$

where $F(t)$ is the fluorescence intensity at 415 nm at various time intervals, $F(0)$ is the fluorescence intensity at 415 nm at $t=0$, and k is the subunit exchange rate constant. The constants D_1 and D_2 were determined using conditions at which $D_1 + D_2 = 1$ at $t=0$ and D_1 is the fluorescence intensity at $t=\infty$. MGO modification alone of α A-crystallin without fluorescent tags did not affect the LYI emission (see Supplementary Data).

RESULTS

Effect of OMIU and MGO Modifications on Chaperone Function of Human α A-Crystallin—We wanted to determine how homoarginine residues, formed by the reaction of lysine residues and OMIU, and the subsequent modification of these residues by MGO influenced the chaperone function of α A-crystallin. We first determined the capacity of unmodified and modified α A-crystallins to prevent DTT-induced insulin aggregation and heat-induced citrate synthase (CS) aggregation. With a ratio of 1:4 (w/w) of unmodified α A-crystallin to insulin, we found ~30% protection against protein aggregation (Fig. 2A), but with a similar ratio of OMIU-modified (0.25 M) α A-crystallin to insulin, protection declined to ~3% (Fig. 2A). Chaperone function was completely lost when the OMIU concentration was increased to 0.5 M or 1.0 M (Fig. 2A).

Surprisingly, subsequent modification of OMIU-treated α A-crystallin with 1 mM MGO not only restored but also markedly improved its chaperone function (Fig. 2A). At a chaperone-to-substrate (insulin) ratio of 1:4, as above, MGO-modified α A-crystallin (no OMIU treatment) provided ~60% protection against insulin-induced aggregation (Fig. 2A). More importantly, MGO strikingly enhanced the chaperone function of OMIU-treated α A-crystallin. This phenomenon was obvious at all concentrations of OMIU (0.25 to 1 M), with an increase

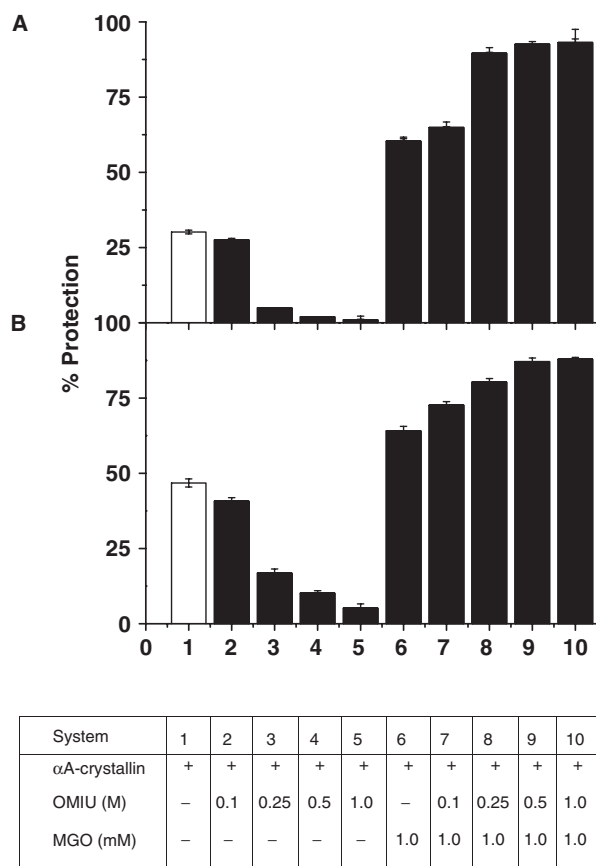


Fig. 2. Chaperone function of OMIU and MGO-modified α A-crystallin. DTT-induced aggregation of 0.32 mg/ml insulin at 25°C (A) and thermal aggregation of 0.06 mg/ml CS at 43°C (B) with and without α A-crystallins. Chaperone function of unmodified α A-crystallin (bar 1) and α A-crystallin incubated with various concentrations of OMIU: 0.1 M (bars 2 and 7), 0.25 M (bars 3 and 8), 0.5 M (bars 4 and 9), 1.0 M (bars 5 and 10) for 48 h at 4°C followed by incubation with 1.0 mM MGO (bars 7 to 10) at 37°C for 48 h. The chaperone function of α A-crystallin incubated with 1.0 mM MGO for 48 h is shown in bar 6. The chaperone:substrate ratio (w/w) was 1:4 and 1:20 for insulin and CS aggregation assays, respectively. Each bar represents the average of three assays.

of ~90% or more beyond that of the OMIU-modified proteins alone (Fig. 2A).

Assays with CS as the substrate gave similar results. At an α A-crystallin:CS (w/w) ratio of 1:20, protection against CS-induced aggregation was 47%. Protection declined if OMIU-modified α A-crystallin was used; the extent of reduction depended on the concentration of OMIU used to modify the protein (Fig. 2B). As in the insulin assay, further modification of OMIU-treated α A-crystallin with MGO not only salvaged the chaperone function, but also markedly improved it. The ability to protect insulin and CS increased ~2-fold in both 0.5 M OMIU + 1.0 mM MGO and 1.0 M OMIU + 1.0 mM MGO α A-crystallin when compared to the unmodified protein (Fig. 2). Data from these assays confirm that MGO modification of OMIU-treated protein not only re-institutes the chaperone function but also augments it beyond the effect of MGO alone. Treatment of BSA with OMIU and MGO as above failed to show similar

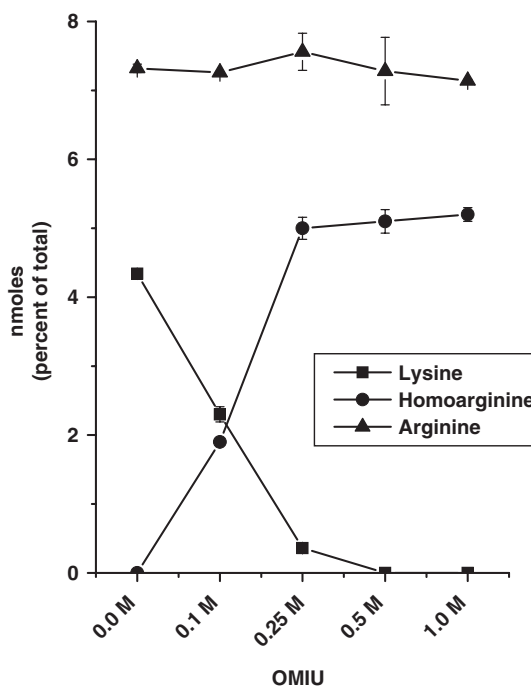


Fig. 3. Content of lysine, arginine and homoarginine in α A-crystallin. α A-crystallin was modified by incubation with 0.1–1.0 M OMIU at 4°C for 48 h. Amino acids in acid-hydrolysed samples were quantified by amino acid analysis. The assay was calibrated with two 5 nM amino acid standards. Data show mean values \pm SD of triplicate measurements.

effects, suggesting that the effects seen with α A-crystallin are specific (data not shown).

We also investigated the effect of OMIU treatment after MGO modification on the chaperone function of α A-crystallin. Treatment with 0.5 M or 1.0 M OMIU after modification with 1.0 mM MGO resulted nearly 70% decrease of chaperone function in the CS aggregation assay and 10–15% decrease in the insulin aggregation assay compared to the unmodified protein (data not shown). These results suggest that guanidination of lysine residues has a superseding effect over MGO modification at the concentrations of OMIU and MGO used in the present study.

Amino Acid Analysis—Amino acid analysis was used to determine the extent of lysine modification during reaction of α A-crystallin with OMIU. Treatment with 0.1 M OMIU converted 50% of lysine residues in α A-crystallin to homoarginine. At higher OMIU concentrations (0.5 M and 1.0 M), all the lysine residues were converted to homoarginine. At these concentrations, OMIU did not modify arginine residues (Fig. 3). In control samples (without OMIU) at pH 10.5, the amino acid content was unchanged, ruling out deamidation of asparagine as a possible effect. In light of our findings on altered chaperone function described above, it would seem that introduction of guanidino groups, not modification of lysines, underlies this change in function.

Identification of Homoargpyrimidine and Argpyrimidine by HPLC—We found that the amount of homoargpyrimidine nearly tripled in 0.5 M OMIU-treated

α A-crystallin (from 1.6 to 4.6 nmol/ μ mol amino acid) compared to the 0.1 M OMIU-treated protein (Fig. 4A). Because all lysines were modified to homoarginine by 0.5 M OMIU (Fig. 3), further increases in the OMIU concentration produced no additional

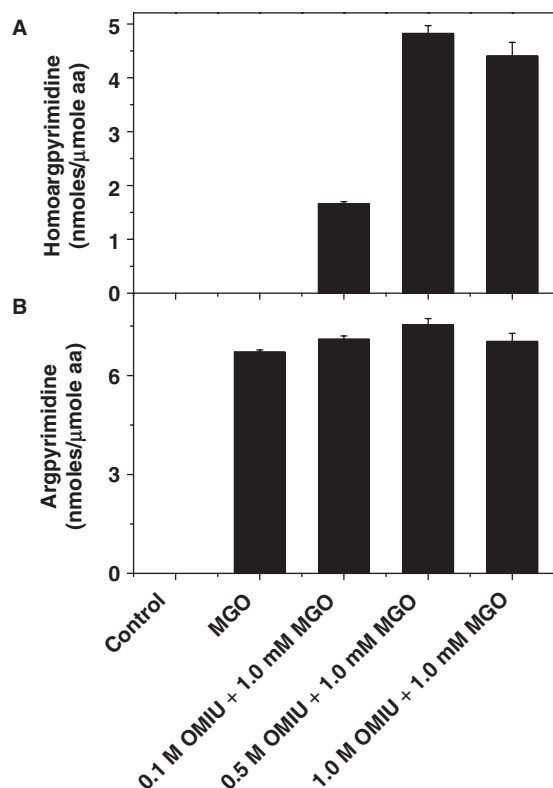


Fig. 4. Homoargpyrimidine and argpyrimidine in modified α A-crystallin. Homoargpyrimidine was identified only in 0.1–1.0 M OMIU modified proteins (4°C for 48 h) and argpyrimidine adduct was identified in MGO (1 mM) and OMIU + MGO modified α A-crystallin (37°C for 48 h). Protein samples (300 μ g each) were hydrolysed in 6 N HCl at 110°C for 20 h and products were measured by reversed phase HPLC with an on-line fluorescence detector. Each bar represents the average of three measurements.

homoargpyrimidine. We also measured argpyrimidine, which is produced by reaction of MGO with arginine residues in α A-crystallin. Treatment with 1.0 mM MGO produced nearly 6 nmol/ μ mol amino acid argpyrimidine; this effect of 1.0 mM MGO was similar across all concentrations of OMIU (Fig. 4B). These data imply that homoarginine is further converted to homoargpyrimidine by MGO treatment. Although homohydroimidazolone could be another major modification produced by MGO, we did not measure this product because of limitations imposed by our HPLC system. However, as shown below we were able to detect it by mass spectrometry.

Homoarginine, Homoargpyrimidine and Homohydroimidazolone Identified by Mass Spectroscopy—To identify homoarginine, homoargpyrimidine and homohydroimidazolone within α A-crystallin, we digested the unmodified protein as well as protein treated with 0.25 M OMIU or 0.25 M OMIU + 1.0 mM MGO with the endopeptidase, Asp-N, and analysed the resulting peptides by LC-ESI-MS/MS. Table 1 summarizes the results. Treatment with 0.25 M OMIU transformed six of the seven lysine residues to homoarginine; the seventh was not detectable under the conditions used. We noted an increase of 42 Da on the modified lysine residues. This agrees with the amino analysis data (Fig. 3), which indicated all lysine residues were modified by 0.25 M OMIU. Further mass increments of 96 Da and 122 Da accrued when MGO (1.0 mM) converted these six homoarginines to homohydroimidazolone and homoargpyrimidine, respectively (Table 1). LC-MS/MS spectra of the peptides in Table 1 confirmed the presence of these modifications. Of the modified residues detected at sites of the original six lysine residues, six were homohydroimidazolone and three were homoargpyrimidine. The peptide 136 DGMLTFCGPKIQTGL $_{150}$ showed both homoargpyrimidine and homohydroimidazolone adducts. Tandem mass spectrum analysis of this peptide confirmed the sites of modification. Figure 5 shows the MS/MS spectra from unmodified, OMIU-modified and OMIU-modified plus MGO-modified α A-crystallin. The results indicate that the mass increase of 42 Da corresponded to the conversion of Lys $_{145}$ to homoarginine, the increase of

Table 1. LC-MS/MS results of Asp-N cleaved unmodified, OMIU-modified and OMIU + MGO-modified α A-crystallin.

No	Sequence	α A-crystallin (Da)	α A-crystallin + OMIU ^a (Da)	α A-crystallin + OMIU + MGO ^b (Da)	Modification Adducts by b
2–23	DVTIQHPWF K R T LGPFPYPSRLF	2704.42 (2704.43) ^c	2746.47 (2746.45)	2800.52 (2800.46)	HHI
69–75	D K FVIFL	880.52 (880.51)	992.54 (992.53)	976.57 (976.54)	HHI
76–83	DV K HFSPE			1053.53 (1053.49)	HHI
				1079.55 (1079.50)	HAP
76–90	DVKHFSPEDLTV K VQ	1740.92 (1740.90)	1782.96 (1782.93)		
	DV K HFSPEDLTV K VQ		1824.99 (1824.95)		
84–90	DLTV K VQ		843.49 (843.48)	897.53 (897.49)	HHI
				923.55 (923.51)	HAP
91–104	DDFVEIHG K HNERQ	1722.80 (1722.81)	1764.86 (1764.83)	1818.92 (1818.84)	HHI
136–150	DGMLTFCG P K I QTGL	1636.84 (1636.80)	1678.85 (1678.82)	1732.87 (1732.83)	HHI
				1758.94 (1758.84)	HAP
35–57	DLLPFLSSTISPYRQSLF R TVL	2715.53 (2715.46)		2795.54 (2795.49)	HAP

HHI, homohydroimidazolone; HAP, homoargpyrimidine.

^aMass increase of 42 Da. ^bMass increase of 96 Da for HHI and 122 Da for HAP in samples modified with 0.25 M OMIU + 1.0 mM MGO.

^cObserved mass and calculated mass in parenthesis.

96Da corresponded to the conversion of homoarginine to homohydroimidazolone and the increase of 122Da corresponded to the conversion of homoarginine to homoargpyrimidine. Together, these data suggest that hydroimidazolones are the dominant modifications induced by MGO in α A-crystallin. Previous studies also showed higher hydroimidazolone concentrations relative to argpyrimidine in MGO-modified proteins (43). We concluded that the increased chaperone function of α A-crystallin results from formation of homohydroimidazolone and homoargpyrimidine adducts within the protein structure.

Effect of OMIU and MGO Modification on Surface Hydrophobicity of α A-Crystallin—The surface hydrophobicity and chaperone function of α -crystallin are strongly

correlated (12, 44–46). Earlier studies showed that structural perturbation influences exposure of hydrophobic residues and consequently, the chaperone function of α -crystallin (24, 45–47). To determine whether the formation of homoarginine, homohydroimidazolone and homoargpyrimidine accompanies altered surface hydrophobicity, we probed unmodified and modified proteins with the hydrophobic probe, 2-(*p*-toluidino) naphthalene-6-sulphonic acid, sodium salt (TNS). The fluorescence intensity of this reagent increases when it binds to hydrophobic regions of a protein (15, 20). Figure 6 shows the intense fluorescence of TNS [emission maximum (λ_{\max}) at 432 nm] when bound to unmodified α A-crystallin. We found that neither the OMIU-modification nor OMIU + MGO modification altered the λ_{\max} . However, the fluorescence

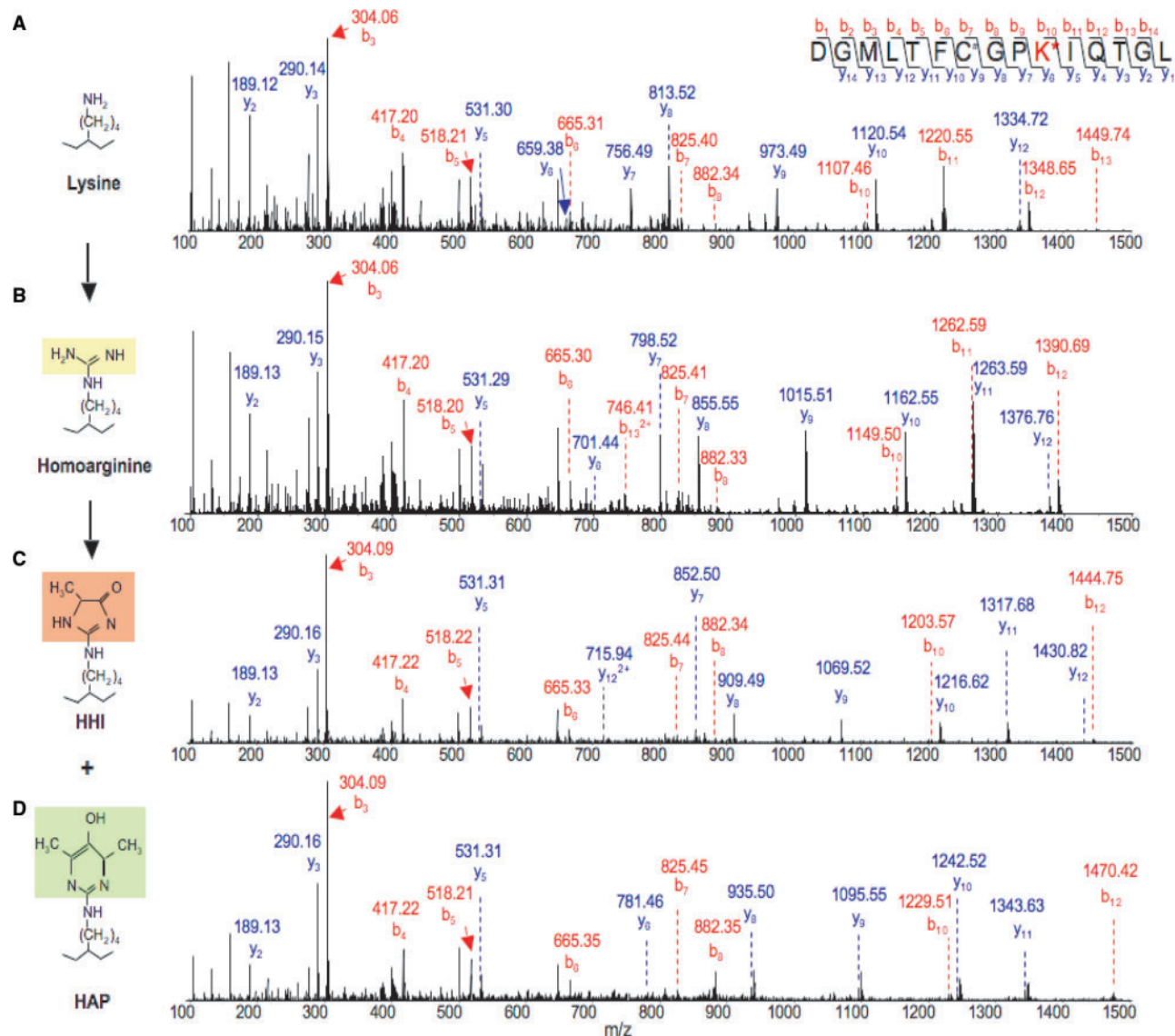


Fig. 5. LC-MS/MS-spectroscopy detection of MGO adducts on OMIU-modified and unmodified α A-crystallin. LC-MS/MS spectra of ${}_{135}\text{DGMLTFCGPKIQTGL}_{150}$ unmodified (A), OMIU-modified (B), OMIU + MGO-modified α A-crystallin (C and D). Cysteines were alkylated by treatment with idoacetamide.

Precursor ions (doubly charged) subjected to MS/MS analyses were m/z 819.42 (A), 840.44 (B), 867.43 (C) and 880.45 (D), respectively. Highlighted portions of structure indicate the adducts that are derived from modifications.

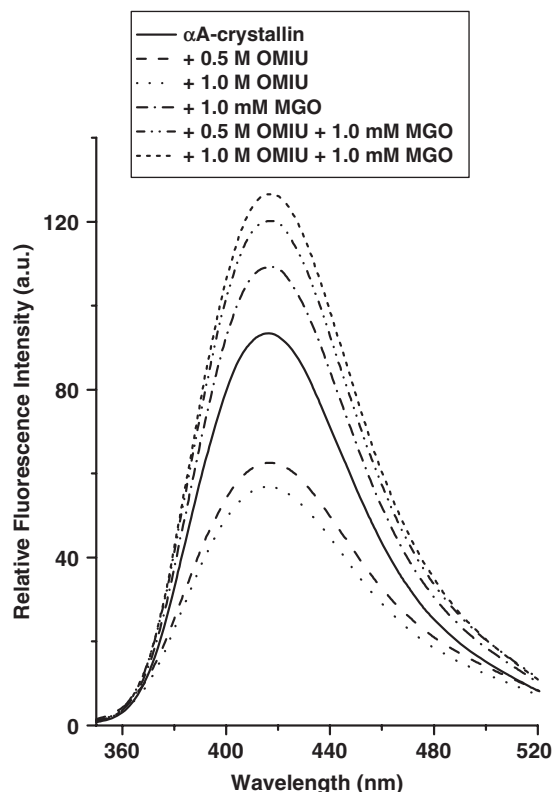


Fig. 6. **Fluorescence spectra of TNS-bound α A-crystallin.** The protein concentration was 0.1 mg/ml; the TNS concentration was 100 μ M. The excitation wavelength was 320 nm. Fluorescence spectra of different samples at 25°C were recorded from 350 to 520 nm at a 0.5 nm interval.

intensity of TNS bound to α A-crystallin modified by 0.5 M or 1.0 M OMIU was reduced by \sim 32% and \sim 38%, respectively, as compared to the unmodified protein. Subsequent modification of these two proteins by MGO (1.0 mM) resulted in an increase in fluorescence intensity (\sim 30% and \sim 39%) above that of unmodified α A-crystallin. We also observed that modification by 1.0 mM MGO alone increased the fluorescence intensity of TNS bound to α A-crystallin by \sim 20% when compared to the unmodified protein. These findings suggest that formation of homo-hydroimidazolone and homoargpyrimidine enhanced the fluorescence intensity and thus the hydrophobicity of the protein. Based on these results, we conclude that the enhanced chaperone function of α A-crystallin with OMIU and MGO modifications results from an increase in surface hydrophobicity.

Effect of OMIU and MGO Modification on the Secondary and Tertiary Structure of α A-Crystallin—Earlier studies indicate that structural perturbation of α -crystallin leads to altered chaperone function (12, 24, 26, 27, 44, 45). To determine if structural perturbation dictated the functional changes in α A-crystallin, we utilized the intrinsic fluorescence (due to the presence of tryptophan residues), near- and far-UV CD spectroscopy. Intrinsic fluorescence spectra indicated significant differences between the unmodified and modified proteins (Fig. 7). The fluorescence emission maximum (λ_{\max}) of unmodified α A-crystallin was 344 nm.

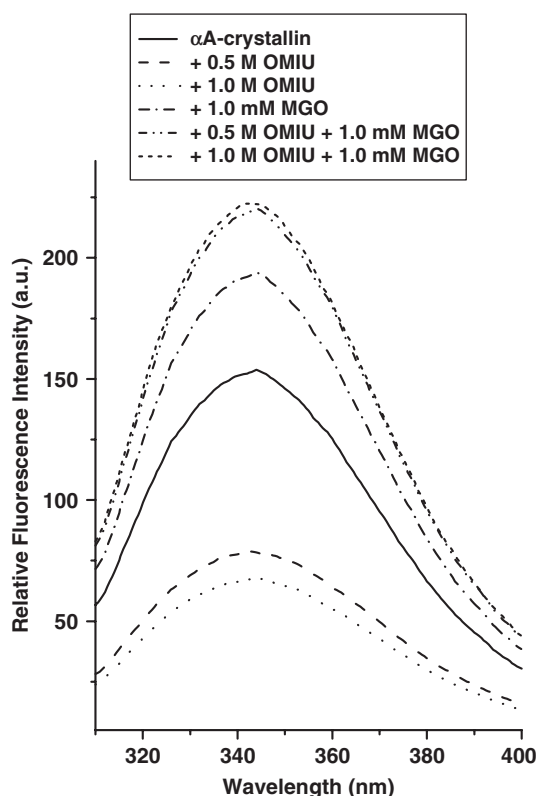


Fig. 7. **Intrinsic fluorescence spectra of OMIU and MGO-modified α A-crystallin.** Fluorescence spectra of unmodified and modified proteins (0.1 mg/ml) were recorded at 25°C from 310 to 400 nm with a LS 55 Perkin Elmer spectrofluorometer (bandwidth of 5 nm for excitation and emission modes). The excitation wavelength was set to 295 nm.

Although λ_{\max} did not change in either OMIU-modified or OMIU+MGO-modified proteins. However, the fluorescence intensity decreased \sim 1.5-fold in proteins modified with 0.5 M and 1.0 M OMIU compared to unmodified α A-crystallin. Modification with MGO increased the fluorescence intensity in all protein preparations. The increase was 28% in the protein modified with 1.0 mM MGO (without OMIU), and \sim 54% in OMIU+MGO modified proteins. These changes in fluorescence intensity suggest that the tryptophan microenvironment is affected by OMIU and MGO treatments.

We also studied perturbation on the tertiary level with near-UV CD spectroscopy (Fig. 8). The near-UV CD spectrum profile obtained for unmodified α A-crystallin is consistent with previously published results (26, 48). The microenvironment of phenylalanine was unaffected by protein modifications introduced by treatment with OMIU or OMIU+MGO. Because differences existed between the unmodified and modified proteins in the region of 270–290 nm (Fig. 8), we assume that the tyrosine and tryptophan microenvironments were perturbed. Together, these results imply that MGO and OMIU treatments perturb the tertiary structure differently. Perturbation by MGO presumably leads to variation in hydrophobic sites on the protein surface that ultimately controls the chaperone function of α A-crystallin.

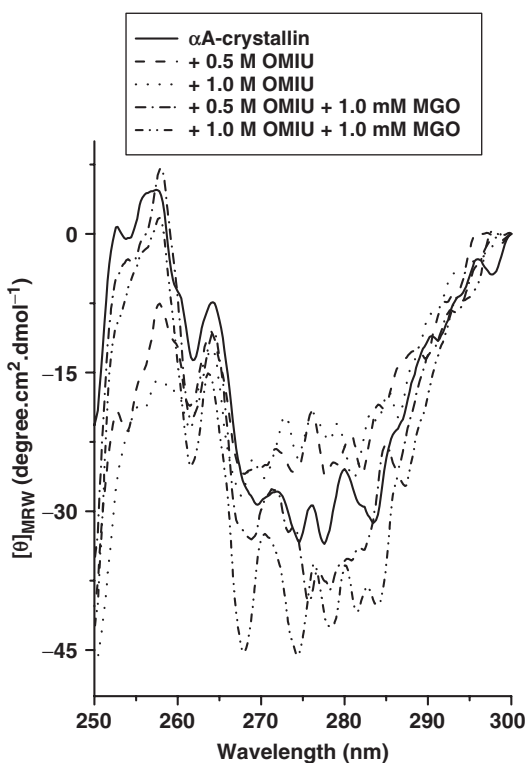


Fig. 8. Near-UV CD spectra of OMIU and MGO-modified α A-crystallin. Spectra were collected using a Jasco J810 spectropolarimeter with a 10 mm path length CD cell. The protein concentration was 1.0 mg/ml. Each spectrum was the average of five scans.

We determined the secondary structure of unmodified, OMIU-modified and OMIU + MGO modified α A-crystallin by far-UV CD spectroscopy (Fig. 9). The CONTINLL program was used to estimate the secondary structural elements (40). We found that β -sheet content of α A-crystallin did not change by modifications (data not shown) and concluded that neither OMIU nor OMIU + MGO markedly perturbed the secondary structure.

Effect of OMIU and MGO Modifications on the Subunit Exchange Rate of α A-Crystallin—Many sHSPs, including α -crystallin, exchange subunits between their oligomers (42, 49–51). We used the fluorescence resonance energy transfer (FRET) technique to monitor the effect of OMIU and OMIU + MGO modifications on subunit exchange of human α A-crystallin. Of the two cysteine residues of human α A-crystallin, Cys₁₃₁ is fully exposed whereas Cys₁₄₂ is completely buried. When we compared the percentage labelling of α A-crystallin with the FRET probes, 4-acetamido-4'-((iodoacetyl)amino)-stilbene-2,2'-disulphonic acid (AIAS) and lucifer yellow iodoacetamide (LYI), we found \sim 1 mol of probe to be covalently bound to 1 mol of α A-crystallin subunit, suggesting covalent modification of the fully exposed Cys₁₃₁ residue. When we mixed 0.4 mg/mL AIAS-labelled and 0.4 mg/ml LYI-labelled proteins at 37°C, we observed a time-dependent decrease in AIAS emission intensity at 415 nm along with a concomitant increase in LYI fluorescence intensity at 525 nm. We plotted the decrease in AIAS fluorescence

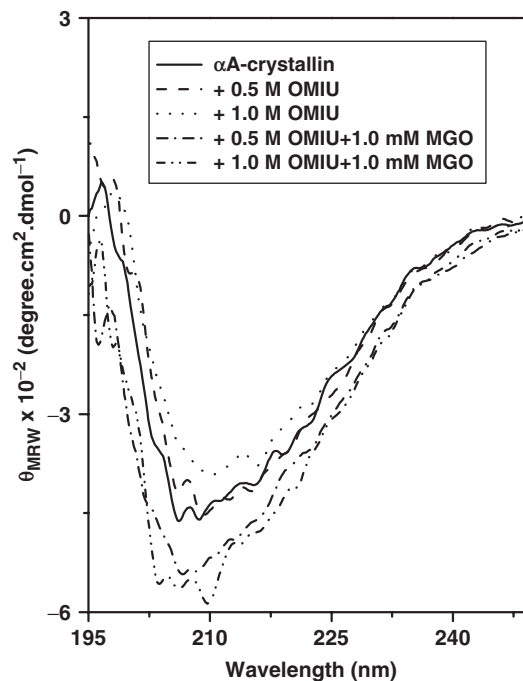


Fig. 9. Far UV CD spectra of OMIU and MGO modified α A-crystallin. Spectra from protein samples (0.2 mg/ml in 10 mM phosphate buffer, pH 7.2) were recorded with a 1.0 mm path length cell. Data interval was 0.2 nm. Each spectrum was the average of five scans.

intensity as a function of time (Fig. 10) to obtain the subunit exchange rate constant (k) according to the equation shown in the MATERIALS AND METHODS. The subunit rate constant for unmodified α A-crystallin was 0.061 min^{-1} , which agrees with previously reported value (42). The k -value for the protein modified by 0.5 M OMIU was 0.032 min^{-1} , which suggests that introduction of a guanidino group reduces the subunit exchange rate by \sim 48%. Subsequent MGO modification, which presumably neutralizes the guanidino groups, restored the subunit exchange rate ($k = 0.063 \text{ min}^{-1}$). Our findings clearly indicate that the introduction of guanidino groups and subsequent neutralization of those groups by MGO have opposing effects on the dynamics of the subunit assembly of α A-crystallin. Several studies suggest that the dynamic properties of subunit assembly in α -crystallin are important for its chaperone activity (20, 42, 52), whereas others have not found such a correlation (53–55). Notwithstanding these differences, our findings indicate a strong correlation between subunit exchange and the chaperone function of α A-crystallin.

DISCUSSION

The purpose of this study was to investigate how the additionally introduced guanidino groups and their reaction with MGO modulate the chaperone function of α A-crystallin. We previously found that modification of discrete arginine residues in α A-crystallin by MGO improved its chaperone ability and that replacement of MGO-modifiable arginine residues by site-directed mutagenesis with neutral alanine duplicated the effects of

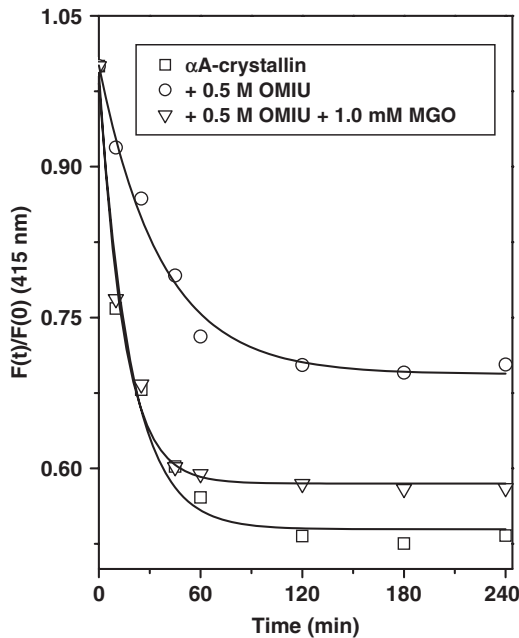


Fig. 10. **Effect of modification on subunit exchange properties of α A-crystallin.** Subunit exchange between AIAS-labelled and LYI-labelled α A-crystallin was measured at 37°C. An equal amount (0.4 mg/ml) of AIAS-labelled and LYI-labelled α A-crystallin was mixed together at 37°C and decrease in relative fluorescence intensity at 415 nm was determined. The subunit exchange rate was calculated using the equation $F(t)/F(0) = D_1 + D_2 e^{-kt}$. The subunit exchange rate constant for unmodified, OMIU and OMIU+MGO modified α A-crystallin was 0.061 min^{-1} , 0.032 min^{-1} and 0.063 min^{-1} , respectively.

MGO (26, 34). These studies led to our hypothesis that the positive charge on the guanidino groups of selected arginine residues controls the chaperone function of α A-crystallin and their removal, either by reaction with MGO or replacement by site-directed mutagenesis, makes α A-crystallin a better chaperone.

Indeed, we found that conversion of lysine residues to homoarginine residues by treatment with OMIU causes α A-crystallin to lose chaperone function, which suggests that introduction of additional guanidino groups is detrimental to the chaperone function. Additionally, this treatment reduced the subunit exchange rate and surface hydrophobicity of the protein. The altered function is not likely due to the removal of positive charge on lysine because previous studies show either no or only partial loss of chaperone function due to chemical modification of lysine residues in α -crystallin (34, 56, 57). Taken together, these observations indicate that introduction of guanidino groups rather than modification of lysine residues causes the decrease in chaperone function.

What we found most remarkable was that treatment with MGO restored the chaperone function of OMIU-treated α A-crystallin. In fact, the gain in chaperone function was consistently greater than that which could be achieved by treating α A-crystallin directly with MGO (without prior OMIU treatment). This strongly suggests that conversion of homoarginine to homohydroimidazolone

and homoargpyrimidine at positions 11, 70, 78, 88, 99 and 145 makes α A-crystallin a stronger chaperone.

Alpha-crystallin has been shown to assist in re-folding of several denatured enzymes *in vitro* (44, 58, 59). Whether the OMIU and OMIU+MGO treatments would decrease or enhance α A-crystallin's ability to assist in such re-folding is currently being investigated.

Based on these findings, we suggest that conversion of lysine to homoarginine reduces the chaperone function and that neutralization of a guanidino group on homoarginine residues after MGO treatment enhances the chaperone function of α A-crystallin. Most lysine residues in α A-crystallin are exposed on the protein surface, so they are accessible to aqueous solvents. Conversion of lysines to more polar homoarginines might result in subtle structural perturbation leading to burial of the hydrophobic pockets responsible for the chaperone function, eventually leading to decreased chaperone function. Establishment of van der Waal's contacts between homoarginines and other amino acids could also restrict substrate binding. Yet another possibility is that protonated guanidino groups on homoarginines could form salt bridges with amino acids, such as glutamic acid and aspartic acid, through guanidinium-carboxylate ion pairs. Whatever the underlying mechanism, decreased accessibility of the substrate-binding hydrophobic regions for target proteins seems to be involved in reducing chaperone function. Loss of chaperone function when α -crystallin is chemically cross-linked also has been reported (57), supporting our notion that decreased flexibility, and consequently a decrease in substrate-binding sites, reduces the chaperone function of OMIU-modified α A-crystallin. That the TNS-binding studies showing reduced surface hydrophobicity reinforces this view. Reaction of the homoarginine guanidino group with MGO might collapse existing salt bridges to expose substrate-binding hydrophobic sites. The end result would be enhanced chaperone function. Our finding of a substantial increase in TNS binding after MGO treatment indicates this possibility.

Our results support the hypothesis that the guanidino group on selected arginine residues in α A-crystallin controls the chaperone function of the protein. If the arginines are chemically modified or removed by site-directed mutagenesis, the protein gains chaperone function. It is debatable whether such a gain of function is beneficial for the lens. The α -crystallin–target protein association in the lens would likely be irreversible because there are no other chaperones to aid in the dissociation of the α -crystallin–substrate complex. A recent study by Koteiche and Mchaourab (60) showed that during chaperoning of proteins, α -crystallin induced unfolding of nascent proteins, causing them to aggregate as well. Thus, even though modification of selected arginine residues by MGO with a gain in chaperone function might seem beneficial, it could prove detrimental to the lens because of the enhanced irreversible α -crystallin–substrate complex formation that could lead to greater light scattering.

This study was supported from National Institutes of Health (NIH) grants R01EY-016219 and R01EY-09912 (RHN), P30EY-11373 (Visual Sciences Research Center

of CWRU), Research to Prevent Blindness (RPB), NY, Ohio Lions Eye Research Foundation and Carl F. Asseff, M.D. Professorship to R.H.N. The authors thank Michael Zagorski and Krzysztof Palczewski of Case Western Reserve University, Cleveland, OH for use of CD spectropolarimeter and the fluorescence spectrofluorometer.

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